

Spring 2022



Chem F325L Organic Chem II Lab Syllabus University of Alaska Fairbanks, Spring 2022

Course Information

Chemistry F325, Organic Chemistry II Laboratory, 1.0 Credits Reichardt 245

Co-Requisites: Lecture component of Chem 325, Organic Chemistry II. **Pre-requisites**: Chem 321 Organic Chemistry I (includes Laboratory)

Section	CRN	Day	Time	Teaching Assistant
F01	34386	Thursday	11:30 2:30	TBA
F02	34387	Thursday	2:45 5:45	TBA
F03	34388	Thursday	6:00 9:00	TBA

Instructor

Thomas Green, Professor of Chemistry Reichardt 174, Phone: Home 907-452-6370 Cell 907-744-2726 <u>Email: tkgreen@alaska.edu</u> Office Hours: Tuesday 1-3 pm, by Zoom

Mode of Delivery: In Person Laboratory.

Teaching Assistant Office Hours

See TA Office Hours. https://www.uaf.edu/chem/clc/

Course Materials Required:

- 1. Laboratory Manual with all experiments and reports forms, supplied to you by instructor.
- 2. Lab notebook for recording experimental data, results and conclusions. The lab notebook will be
- supplied by the department. Student Lab Notebook, 2012 Book Factory, Lab-050-7GSS, 50 pages.
- 3. Textbook: Making the Connections³; A How-to-Guide for Organic Chemistry Lab Techniques, 3rd edition, Anne B. Padias, 2015, Hayden McNeil.

Course Description: A laboratory designed to illustrate modern techniques of isolation, purification, analysis and structure determination of covalent, principally organic, compounds. Lab portion will include an introduction to synthetic techniques and spectroscopy. Special fees apply.

Course Goals. Learn the following practical aspects of organic synthesis.

- 1. Common safety procedures.
- 2. Reaction methods
- 3. Isolation Procedures
- 4. Purification techniques
- 5. Spectroscopic and chromatographic analyses
- 6. Introduction to computational methods in chemistry.

Student Learning Outcomes

- 1. Know the hazards associated with common chemicals, especially those encountered in the experiments.
- 2. Know how to safely assemble reaction systems using glassware commonly employed in the organic laboratory. These methods include reflux, heating and cooling of reactions, and addition of reagents.
- 3. Know how to isolate and purify organic products using methods such as extraction, filtration, crystallization, distillation, solvent removal, and thin layer chromatography.
- 4. Learn the importance of stoichiometry to a chemical reaction. Learn how to assess the efficiency of a chemical reaction (percent yield and atom economy).
- 5. Learn the practical aspects of spectroscopic analyses of organic compounds.
- 6. Learn how to build and optimize simple molecules using WebMO/Gaussian and how to measure properties of those molecules.

Instructional Methods

- 1. The instructor or teaching assistant will provide a brief introduction on the practical aspects of organic chemistry, using a combination of Power Point slides and Chalkboard. The Lab Schedule will be available on Blackboard and at the end of this syllabus.
- 2. Laboratory sessions will consist of conducting reactions of organic compounds and their isolation, purification and characterization.
- 3. E O k h -lab and Post-lab components. The Pre-lab portion must be completed prior to coming to lab. If it is not completed, you will not be allowed to work in the lab for that day Your TA will need to verify with her/his initials that you have completed the pre-lab questions. Students are also required to keep a laboratory notebook. The lab notebook will be collected at the midterm, evaluated but not graded, and returned with suggestions for improvement. The lab notebook will be graded at the end of the semester.
- 4. A lab textbook **Making the Connections³** by Anne Padias which describes techniques, glassware, lab notebooks, spectroscopic techniques, etc. Readings will be assigned to the student for each experiment.

Lab Notebook Guidelines:

Before each lab, you should enter the following in the notebook (with pen).

- 1. Title of Experiment
- 2. Hypothesis or Goal of Experiment
- 3.

During lab, you should enter the following,

6. Data and observations. Record actual amounts (volumes or mass) used for each reagent. Record physical constants such as melting point range of the product. If you ran a TLC plate, sketch plate in the notei

Lab Schedule see Canvas for specific Experimental Procedures and Report Forms.

Experiment	Date	Concepts/Techniques	Chapter Wade
No Lab	Jan 13. 20	No Lab	
HW 1: NMR of Unknowns (30)	Jan 27	13C, 1H NMR, IR Structure Determination	13
HW 2: Mass Spectrometry (30)	Feb 3	Structure Determination, Fragmentation of Functional Groups	12
Exp 1: Solvent extraction of Natural Product (50)	Feb 10	Mass Spectrometry; Solvent Extraction	12
Exp 2: Diels-Alder Reaction (50	Feb 17	Reflux, NMR Coupling Constants, Modeling	15

Exp 3a: Iodination of Vanillin

COVID-19.

Students should keep up-toregularly checking this website: <u>https://sites.google.com/alaska.edu/coronavirus/uaf/uaf-students?authuser=0</u> F